

Cambridge Assessment International Education

Cambridge International Advanced Subsidiary and Advanced Level

CHEMISTRY 9701/31

Paper 3 Advanced Practical Skills 1

May/June 2019

CONFIDENTIAL INSTRUCTIONS



This document gives details of how to prepare for and administer the practical exam.

The information in this document and the identity of any materials supplied by Cambridge International are confidential and must NOT reach candidates either directly or indirectly.

The supervisor must complete the report at the end of this document and return it with the scripts.

If you have any queries regarding these confidential instructions, contact Cambridge International stating the centre number, the syllabus and component number and the nature of the query.

email info@cambridgeinternational.org

phone +44 1223 553554 fax +44 1223 553558

This document consists of **7** printed pages and **1** blank page.



2

General information about practical exams

Centres must follow the guidance on science practical exams given in the Cambridge Handbook.

Safety

Supervisors must follow national and local regulations relating to safety and first aid.

Only those procedures described in the question paper should be attempted.

Supervisors must inform candidates that materials and apparatus used in the exam should be treated with caution. Suitable eye protection should be used where necessary.

The following hazard codes are used in these confidential instructions, where relevant:

C corrosive
 HH health hazard
 F flammable
 MH moderate hazard acutely toxic
 O oxidising

N hazardous to the aquatic environment

Hazard data sheets relating to substances used in this exam should be available from your chemical supplier.

Before the exam

- The packets containing the question papers must not be opened before the exam.
- It is assumed that standard school laboratory facilities, as indicated in the *Guide to Planning Practical Science*, will be available.
- Spare materials and apparatus for the tasks set must be available for candidates, if required.

During the exam

- It must be made clear to candidates at the start of the exam that they may request spare materials and apparatus for the tasks set.
- Where specified, the supervisor must perform the experiments and record the results as instructed.
 This must be done out of sight of the candidates, using the same materials and apparatus as the candidates.
- Any assistance provided to candidates must be recorded in the supervisor's report.
- If any materials or apparatus need to be replaced, for example, in the event of breakage or loss, this must be recorded in the supervisor's report.

After the exam

- The supervisor must complete a report for each practical session held and each laboratory used.
- Each packet of scripts returned to Cambridge International must contain the following items:
 - the scripts of the candidates specified on the barcode label provided
 - the supervisor's results relevant to these candidates
 - the supervisor's reports relevant to these candidates
 - seating plans for each practical session, referring to each candidate by candidate number

the attendance register.

© UCLES 2019 9701/31/CI/M/J/19

3

Specific information for this practical exam

During the exam, the supervisor (NOT the invigilator) must do all the experiments and record the results on a spare copy of the question paper, clearly labelled 'supervisor's results'.

If chemicals are prepared in more than one batch, clearly labelled supervisor's results must be provided for each batch. The candidates using each batch must be listed on the supervisor's report.

Apparatus

The apparatus listed must be provided to each candidate.

- 1 × 250 cm³ measuring cylinder
- 2 × stand, clamp and boss
- $1 \times 250 \, \text{cm}^3$ side-arm conical flask, labelled **X**, with bung and approximately 50 cm of plastic/rubber delivery tube to fit **or**
- $1 \times 150 \, \text{cm}^3$ or $250 \, \text{cm}^3$ conical flask, labelled **X**, with 1-hole bung connected to approximately $50 \, \text{cm}$ of plastic/rubber delivery tube to fit
- 1 × tub suitable for acting as trough, minimum capacity 1 dm³
- 1 × crucible, capacity approximately 15 cm³, with lid
- 1 × spatula
- 1 × pipe-clay triangle
- 1 × tripod
- 1 × stop-clock or sight of clock with seconds display
- 1 × crucible tongs
- 8 × test-tube*
- 1 × test-tube rack
- 2 × teat/dropping pipette
- 1 × Bunsen burner
- 1 × heatproof mat
- 1 × wash bottle of distilled water
- 1 × pen (for labelling glassware)

access to balance weighing to a minimum accuracy of 0.1 g

paper towels

red and blue litmus papers

aluminium foil for testing nitrate/nitrite

wooden splints

the apparatus normally used in the centre for use with limewater in testing for carbon dioxide

Where balance provision is limited, some candidates should be instructed to start the exam with different questions. See the current syllabus for balance: candidate ratio.

^{*}Candidates are expected to rinse and reuse test-tubes where possible. Additional tubes should be available.

4

The materials listed in the table must be provided to each candidate.

Materials

Warning: small amounts of SO2 [C][T], which can cause respiratory distress in some people, may be produced. The laboratory must be well ventilated.

label	per candidate	identity	notes
FA 1 [MH]	50 cm³	4.00 mol dm ⁻³ hydrochloric acid	Dilute 340 cm³ of concentrated (35–37%; approximately 11 moldm⁻³) HC1 [C][MH] to 1 dm³.
FA 2	0.6 ± 0.1g	small marble chips (approx. 2–4 mm)	Provide 0.6 \pm 0.1 g of CaCO $_3$ (as small marble chips) in a stoppered container.
FA 3	1.5 ± 0.1g	1.5 ± 0.1 g magnesium carbonate	Provide 1.5 \pm 0.1 g of any powdered form of MgCO $_3$ or MgCO $_3$.Mg(OH) $_2$ in a stoppered container.
FA 4	15cm³	0.2 mol dm-3 potassium iodide	Dissolve $33.2\mathrm{g}$ of KI in each dm $^{\mathrm{3}}$ of solution.
FA 5	15 cm³	0.2 mol dm ⁻³ sodium thiosulfate	Dissolve 49.6g of $Na_2S_2O_3.5H_2O$ in each dm^3 of solution.
FA 6	10 cm³	0.2 mol dm-3 calcium chloride	Dissolve 22.2g of $CaCl_2$ [MH] in each dm 3 of solution.
FA 7	10 cm³	0.5 mol dm ⁻³ magnesium sulfate	Dissolve 123.2g of MgSO $_4$.7H $_2$ O in each dm 3 of solution.
aqueous chlorine [MH][N]	5cm³	0.2 mol dm ⁻³ sodium chlorate(I) (may be labelled sodium hypochlorite)	Dilute 200 cm³ of 1.0 mol dm⁻³ NaClO [C][N] to 1 dm³. This is approximately 3% w/w available chlorine from chlorine-bleaching solutions (may be labelled sodium hypochlorite). Provide in a stoppered container.
aqueous copper(II) sulfate [C][N]	5cm³	0.2 mol dm-3 copper(II) sulfate	Dissolve 49.9g of $CuSO_4.5H_2O$ [C][MH][N] in each dm^3 of solution.

label	per candidate	identity	notes
dilute hydrochloric acid	10 cm³	$2.0\mathrm{moldm^{-3}HC}$	
dilute nitric acid [C]	10 cm³	2.0 mol dm ⁻³ HNO ₃	
dilute sulfuric acid [MH]	10 cm³	$1.0\mathrm{moldm^{-3}H_2SO_4}$	
aqueous ammonia [C][MH][N]	10 cm³	2.0 mol dm ⁻³ NH ₃	See preparation instructions in the current syllabus.
aqueous sodium hydroxide [C]	10 cm³	2.0 mol dm ⁻³ NaOH	If necessary, each of these reagents can be provided as a
aqueous barium chloride or aqueous barium nitrate	10 cm³	$0.1 \text{mol dm}^{-3} \text{BaC} l_2$ or $0.1 \text{mol dm}^{-3} \text{Ba(NO}_3)_2$	I confinition a supply for groups of up to a carrollates. Invigilators must be alert to the risk of contamination and the opportunity for malpractice when using a communal supply.
limewater [MH]	10 cm³	saturated aqueous calcium hydroxide, Ca(OH) ₂	
aqueous silver nitrate	10 cm³	0.05 moldm ⁻³ AgNO ₃	
aqueous acidified potassium manganate(VII) [MH]	10 cm³	$0.01 \text{mol dm}^{-3} \text{KMnO}_4 \text{in}$ $0.5 \text{mol dm}^{-3} \text{H}_2 \text{SO}_4$	

An excess of at least 10% of each material must be prepared to cover accidental loss.

All solutions must be thoroughly mixed.

Materials must be labelled only as specified in the 'label' column. The identities of chemicals labelled with letter codes, e.g. FA 1, may be different from their descriptions in the question paper. Candidates must use the descriptions given in the question paper.

If you are unable to source any of these chemicals, you must contact Cambridge International as far as possible in advance of the exam for

Buy IGCSE, O / A Level Books, Past Papers & Revision Resources Online on Discounted Prices Visit: www.TeachifyMe.com / Shop Call / WhatsApp: (0331-9977798)

BLANK PAGE

Permission to reproduce items where third-party owned material protected by copyright is included has been sought and cleared where possible. Every reasonable effort has been made by the publisher (UCLES) to trace copyright holders, but if any items requiring clearance have unwittingly been included, the publisher will be pleased to make amends at the earliest possible opportunity.

To avoid the issue of disclosure of answer-related information to candidates, all copyright acknowledgements are reproduced online in the Cambridge Assessment International Education Copyright Acknowledgements Booklet. This is produced for each series of examinations and is freely available to download at www.cambridgeinternational.org after the live examination series.

Cambridge Assessment International Education is part of the Cambridge Assessment Group. Cambridge Assessment is the brand name of the University of Cambridge Local Examinations Syndicate (UCLES), which itself is a department of the University of Cambridge.

© UCLES 2019 9701/31/CI/M/J/19

Buy IGCSE, O / A Level Books, Past Papers & Revision Resources Online on Discounted Prices Visit: www.TeachifyMe.com / Shop Call / WhatsApp: (0331-9977798)

7

Supervisor's report

Syllabus and component number			/		
Centre number					
Centre name	 	 		 	
Time of the practical session	 	 		 	
Laboratory name/number	 	 		 	

Give details of any difficulties experienced by the centre or by candidates (include the relevant candidate names and candidate numbers).

You must include:

- any difficulties experienced by the centre in the preparation of materials
- any difficulties experienced by candidates, e.g. due to faulty materials or apparatus
- any specific assistance given to candidates.

Buy IGCSE, O / A Level Books, Past Papers & Revision Resources Online on Discounted Prices Visit: www.TeachifyMe.com / Shop Call / WhatsApp: (0331-9977798) 8
If chemicals have been prepared in more than one batch, list the candidates using each batch.
Declaration
1 Each packet that I am returning to Cambridge International contains the following items:
the scripts of the candidates specified on the barcode label provided
the supervisor's results relevant to these candidates
the supervisor's reports relevant to these candidates
seating plans for each practical session, referring to each candidate by candidate numberthe attendance register
Where the practical exam has taken place in more than one practical session, I have clearly labelled the supervisor's results, supervisor's reports and seating plans with the time and laboratory name/number for each practical session.
3 I have included details of difficulties relating to each practical session experienced by the centre or by candidates.
4 I have reported any other adverse circumstances affecting candidates, e.g. illness, bereavement or temporary injury, directly to Cambridge International on a <i>special consideration form</i> .
Signed (supervisor)
Name (in block capitals)

© UCLES 2019 9701/31/CI/M/J/19